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Types Of Reactions

Practice your six types of reaction.

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Types of chemical reaction | The
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Six Types of Chemical Reaction

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Worksheet Answers Balance the following reactions and indicate which of the six types of chemical reaction are being represented: 1) $2 \text{NaBr} + 1 \text{Ca(OH)}_2 \rightarrow 1 \text{CaBr}_2 + 2 \text{NaOH}$ Type of reaction: double displacement 2) $2 \text{NH}_3 + 1 \text{H}_2\text{SO}_4 \rightarrow 1 (\text{NH}_4)_2\text{SO}_4$ Type of reaction: synthesis 3) $4 \text{C}_5\text{H}_9\text{O} +$

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Answers
27 O₂ + 20 C₂H₆ → 40 CO₂ + 18 H₂O Type of reaction: combustion

Six types of chemical reaction
worksheet

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Answers Using your knowledge of the

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Types Of Reactions

Six types of chemical reaction, balance the following equations and indicate which of the six types of reaction they are:

1) $2 \text{LiH} + 1 \text{F}_2 \rightarrow 2 \text{LiF} + 1 \text{H}_2$ Type of reaction: Single replacement

2) $2 \text{Fe}(\text{OH})_2 \rightarrow 1 \text{Fe}_2\text{O}_3 + 3 \text{H}_2\text{O}$ Type of reaction: Decomposition

3) $3 \text{H}_2\text{SO}_4 + 2 \text{Ga}(\text{OH})_3 \rightarrow 6 \text{H}_2\text{O} + 1 \text{Ga}_2(\text{SO}_4)_3$

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Type of reaction: Acid-base

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Reserved Types of Reactions

Worksheet □ Solutions 1) $3 \text{ NaBr} + 1 \text{ H}$

$3 \text{ PO}_4 + 1 \text{ Na}_3 \text{ PO}_4 + 3 \text{ HBr}$ Reaction

type: _____ double displacement 2) 3

$\text{Ca(OH)}_2 + 1 \text{ Al}_2 (\text{SO}_4)_3 \rightarrow 3 \text{ CaSO}_4$

$+ 2 \text{ Al(OH)}_3$

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Answers
types of chemical reactions are:

Combination reaction; Decomposition reaction; Displacement reaction; Double Displacement reaction; Precipitation

Types Of Chemical Reactions

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Another very common chemical
reaction is of two types, i.e. single

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Types Of Reactions

displacement and double displacement. In single displacement reaction, any one chemical partner exchanges from reactants to products while two sets of chemical partners exchanges from reactants to products. An example for single displacement reaction is $XY + Z \rightarrow XZ + Y$.

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Updated 2-10-16. Types of reaction
and reaction products: Overview of
chemical reactions: Some info from a
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Types Of Reactions

reactions. Pretty much covers the stuff above. Types of reaction: There are several categories into which chemical reactions can be placed, and knowing them can help you to figure out how to predict the ways in which chemical reactions will operate.

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Types of Reactions Worksheet

Balance the following equations and
indicate the type of reaction taking
place: $\text{H}_3\text{P}_04 + \text{HBr} \rightarrow \text{Na}_3\text{P}_04 + \text{NaBr} +$
 $\text{A}_{12}(\text{S}_04)_3$ Type of reaction: $\text{CaS}_04 +$

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Types Of Reactions

Approved
Type of reaction: $\text{Mg} + \text{C}_2\text{H}_4 + \text{PbSO}_4$
 $\text{NH}_3 + \text{H}_2\text{O} + \text{O}_2 \text{PbSO}_3 + \text{MgO} \text{Fe} +$
Type of reaction: $\text{H}_2\text{O} \text{CO}_2 +$ Type of
reaction: Type of reaction: H_2 Type of
reaction: $12 \text{SO}_3 \text{N}_2 + \text{H}_2\text{SO}_4$

* $\text{LiNaNO}_3 + \text{PbSO}_4 \text{C}_3\text{H}_8 + \text{CO}_2$ 1-120

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Answers
Zn + 2HCl → ZnCl₂ + H₂ 10. 11. 12. 13 ...

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Reaction Products - Solutions Balance the equations and predict the products for the following reactions: 1) $3 \text{Na} + 1 \text{FeBr}_3 \rightarrow 3 \text{NaBr} + 1 \text{Fe}$ 2) $2 \text{NaOH} + 1 \text{H}_2\text{SO}_4 \rightarrow 1 \text{Na}_2\text{SO}_4 + 2 \text{H}_2\text{O}$

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