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At that point, the reaction had finished. 2 a i Graph showing amount of product increasing with time. ii.The graph showing the amount of reactants with time. b Answer The gradient of the line on the graph of concentration versus time give the information of the rate of a reaction.

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AQA GCSE Chemistry C13 The Earth ' s atmosphere Kerboodle Answers Summary Questions: Page No. 204. 1. Answer. Over time there is an increase in the percentage of nitrogen from 5% to 75%. On the other hand there is decrease in percentage of oxygen from 40% to 25%, methane from 30% to trace and ammonia from 25% to trace.

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A coffee-cup calorimeter contains 50.0 mL of distilled water at 22.7 ° C. Solid ammonium bromide (3.14 g) is added and the solution is stirred, giving a final temperature of 20.3 ° C. Using the same assumptions as in Example 9, find H soln for NH 4 Br (in kilojoules per mole). Answer: 16.6 kJ/mol